2013 Vol. 15, No. 1 148–151

Catalytic Three-Component One-Pot Reaction of Hydrazones, Dihaloarenes, and Amines

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ABSTRACT



A new three-component assembly reaction between *N*-tosylhydrazones, dihalogenated arenes, and various primary and secondary amines was devised, producing nitrogen-containing 1,1'-diarylethylenes in good yields. The two C-C and C-N bonds formed through this coupling have been catalyzed by a single Pd-catalyst in a one-pot fashion.

The search for practically simple and highly efficient methods to access complex structures is an ongoing challenge for the organic chemist. In this context, transition-metal-catalyzed multicomponent reactions (MCRs), most notably using palladium, allow the efficient construction of elaborate molecules from simple precursors, in a single step without isolation of any intermediate. In these types of reactions, various bonds can be successively created in one operation, and it is of great interest when a single catalyst is able to initiate two or more mechanistically distinct reactions in a multicomponent transformation (autotandem catalysis). Such a process is particularly attractive in view of ecological and economical concerns (e.g., catalyst and solvent economy, operational simplicity, etc.).

In recent years, *N*-tosylhydrazones have attracted extensive attention because of their various useful applications in organic synthesis.³ In particular, they are

valuable and readily available reagents in C-C⁴ and C-heteroatom⁵ bond-forming reactions through metal-catalyzed processes. Because there are many different reactions that can be catalyzed by the same catalyst, great potential exists to link these reactions in a sequence. Recently, *N*-tosylhydrazones containing an amino group have been coupled with *o*-dihalobenzenes⁶ through an efficient sequential C=C bond formation and intramolecular C-N cross-coupling to produce phenanthridine and acridine derivatives.

Although significant progress has been achieved on sequential transformations of *N*-tosylhydrazones through autotandem catalysis, ⁷ to our knowledge, only one study

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Scheme 1. MCRs of Tosylhydrazones with Dihaloarenes and Amines through Palladium-Catalyzed Carbene Migratory Insertion and C-N Cross-Coupling

has explored MCRs of *N*-tosylhydrazones with an aryl halide and a terminal alkyne.⁸

As a continuation of our interest in the 1,1-diarylethylene unit synthesis, as a promising cytotoxic agent, a combined with the development of MCRs, 11 we report here a novel Pd-catalyzed three-component coupling between N-tosylhydrazones 1, dihaloarenes 2, and amines 3 leading to a faster C=C bond formation and an efficient intermolecular C—N cross-coupling (Scheme 1). The value of this MCR is in its brevity and ease of synthesis of the desired substrates 4, as well as in the palladium-based catalytic system preventing both the formation of symmetrical disubstituted byproducts from dihaloarenes 2 and reaction of amines 3 with N-tosylhydrazones 1.5b In addition, the availability of amines and dihaloarenes makes this approach sufficiently diversity-oriented, thus fulfilling the recent demand for the generation of large combinatorial chemical libraries.

We began the exploration of this MCR process with a model reaction between *N*-tosylhydrazone **1a** (1.2 equiv), 4-chloroiodobenzene **2a** (1 equiv), and *p*-anisidine **3a** (1 equiv) using our previously optimized protocol: ¹² PdCl₂-(MeCN)₂ (2 mol %), Xphos (4 mol %), and NaOtBu (3 equiv) in fluorobenzene (Table 1). Under these conditions, the desired product **4a**, formed through two consecutive processes, C–C cross-coupling and intermolecular C–N bond forming reactions, was isolated in a moderate 47% yield after 14 h at 110 °C. Careful analysis by NMR showed a significant amount of C—C monocoupling

compound **5a**, together with symmetrically disubstituted byproduct **6a** (entry 1). Products derived from undesired side reactions, such as mono- or diamination of **2a**, were not observed, ¹³ clearly suggesting that the C=C bond formation occurred faster than C—N cross-coupling. Examination of reaction parameters revealed that increasing the amount of base to 3.5 equiv slightly raised the yield of **4a** (entry 2). We were delighted to find that the use of a 1/1 ratio of *p*-anisidine **3a** and hydrazone **1a** leads to improvement in performance of this MCR, and **4a** was isolated in a 72% yield (entry 3). Finally, fine-tuning of the temperature and the reaction time led to **4a** in an excellent isolated yield of 92% (entry 5), showing that this MCR is viable and indeed highly selective.

Table 1. Optimization of the One-Pot Reaction Conditions^a

entry	aniline (equiv)	time (h)/ temp °C	4a	5a	6a	yield of $\mathbf{4a} (\%)^c$
1	1.0	14/110	62	33	5	47
2	1.0	14/110	72	17	11	55^d
3	1.2	14/110	92	0	8	72^d
4	1.2	6/110	94	0	6	75^d
5	1.2	6/120	100	0	0^e	$92^{d,f}$

^aThe reactions were carried out in a sealed tube with hydrazone 1a (1.2 mmol), 2a (1 mmol), 3a (x equiv), PdCl₂(MeCN)₂ (2 mol %), Xphos (4 mol %), and NaO/Bu (3 equiv) at mentioned temperature in PhF (4.0 mL). ^b Ratio was determined by ¹H NMR in the crude mixture; see Supporting Information. ^c Yield of isolated product 4a. ^d 3.5 equiv of NaO/Bu were used. ^e Not detected by ¹H NMR in the crude reaction. ^f Using Pd₂(dba)₃, Pd(OAc)₂, or Pd(acac)₂ in place of PdCl₂(MeCN)₂ furnished 4a in 51%, 78%, and 69% yield, respectively.

Next, the substrate diversity and the efficiency of the PdCl₂(MeCN)₂/Xphos catalytic system on this three-component coupling were examined with various hydrazone, dihaloarene, and amine components (Table 2). Initially, the scope of the reaction was examined with respect to the dihalogenated aromatic system and the nature of amine, including primary and secondary anilines as well as aliphatic amines. Reactions with *para-*, *meta-*, and *ortho*-anisidines 3a-c proceeded efficiently to form the expected products 4a-c in good yields (entries 1–3). The electronic nature of the substituents on the anilines did not have a significant effect on the reaction. Electron-donating or -withdrawing substrates all reacted to give the

Org. Lett., Vol. 15, No. 1, 2013

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Table 2. Scope of One-Pot, Three-Component Reaction of N-Tosylhydrazones with Dihalogenated Arenes and Amines^a

entry	1	2	3	compound 4	yield (%) ^b	entry	1	2	3	compound 4	yield (%) ^b
1	OMe NNHTs OMe 1a	2a	H ₂ N 3a	OMe OMe 4a	92	15	1a	2e	3i	OMe N	72
2	la	2a	H_2N OMe $3b$	Meo OMe N OMe 4b	76	16		7.5	2.	4o	51
3	1a	2a	H ₂ N OMe 3c	Meo OMe N OMe	77	16	1a	2f	3i	MeO OMe N 4p	51
4	la	2c	3a	4c OMe OMe	82	17	1a	2b	H ₂ N- <i>n</i> -C ₈ H ₁₇ 3k	MeO MeO	50
				4d OMe		18	1a	2a	HN(n-Bu) ₂ 31	MeO OMe N(r-Bu),	57
5	la	2d	3a	MeO OMe 4e	73°	19	1a	2a	HN N. Me	MeO OMe N	72
6	1a	2a	H ₂ N Me 3d	Me Me	71				3111	4s	,
7	1a	2a	. J	4f	67	20	la	2a	HN=CPh _z 3n	MeO OMe NH ₂ 4t	45 ^d
			3e	4g		21 Me	NNHTs OMe 1	2a b	3a	Meo Meo 411	88
8	1a	2a	3f	Meo OMe N	65	22	NNHTS	2a	3i	G C N	56
9	1a	2g	3a	MeO OMe OMe	70	23	1c NNHTs	2b	3a	4v OMe	35
10	1a	2a	HN	MeO OMe N	55		1d			4w	
10	Ia	2 a	3g	4j	33	24 ,	MeO 1e	2a	3i	MeO 4x	72
11	1a	2a	HN Me	MeO NMe NMe	72	25	MeO If	2a	3i	Meo Charles	93
12	1a	2a	3h	4k	85	26 Me	PO NNHTs	2a	3i	4y MeO	90
			31	4I		Me	1 1		31	MeO OMe 4z	70
13	1a	2e	3i	MeO OMe 4m	81	27 Ma	I INNIIIS	2a	3a	MeO OMe	75
14	1a	2a	HN 3j	Meo CoMe N	78	Me	OMe 1	h		оме H 4аа	
			٧,	4n		28	li	2a	3i	4ab	68

^a The reactions were carried out in a sealed tube with hydrazones **1** (1.2 mmol), aryl halides **2** (1 mmol), amines **3** (1.2 mmol), PdCl₂(MeCN)₂ (2 mol %), Xphos (4 mol %), and NaO*t*Bu (3.5 equiv) at 120 °C in PhF (4.0 mL). ^b Isolated yield of product **4**. ^c Aniline **3a** was added after 1 h of coupling between **1a** and **2d**. ^d**4r** was obtained by acidic hydrolysis of *N*-aryl imine adduct in a THF/HCl solution; see Supporting Information.

corresponding products 4a-g in good yields (entries 1-7). In the same manner, switching the halide group from the *para*-position to the *meta*- or *ortho*-position on the aryl ring of 2 does not significantly influence the overall yield of the reaction, demonstrating that steric factors have little influence on the rate of C=C and C-N bond formation

(compare entries 1 and 4, 5). Interestingly, this three-component coupling was found also to proceed successfully with heterocyclic amines, such as 5-aminoquinoline **3f**, furnishing **4h** in a 65% yield (entry 8). To illustrate the usefulness of this MCR, we prepared from hydrazone **1a** and dihalogenated aromatic compound **2g** the

Org. Lett., Vol. 15, No. 1, 2013

1,1-diarylethylene **4i** in one step, which could be regarded as an analogue of antivascular *iso*NH₂-combretastatin-A4¹⁴ (entry 9).

Next we examined the efficiency of this MCR with secondary anilines. Thus, reactions with diphenylamine 3g, N-methylaniline 3h, indoline 3i, and 1,2,3,4tetrahydroquinoline 3i all reacted to give the corresponding coupling products with yields ranging from 55 to 85% (entries 10-14). It is intersting to note that this reaction can be applied successfully to a trihalogenated aromatic system. A good yield of 40 was obtained when using 1,3dichloro-5-iodobenzene 2e as a coupling partner. Accordingly, one C=C bond and two C-N bonds were formed efficiently, representing an average 90% yield for each of the three steps of this MCR (entry 15). In addition, we found also that our coupling conditions proved also to be successful with dihalogenated heteroaromatic partner 2f; in this case, compound 4p was obtained in a satisfactory 51% yield (entry 16).

Furthermore, we investigated whether it might be possible to carry out this MCR with aliphatic amines. To our delight, satisfactory to good yields were obtained with primary and secondary amines, such as *n*-octylamine **3k**, dibutylamine **3l**, and 1-methylpiperazine **3m** (entries 17–19). Extending this MCR to include benzophenone imine **3n** as an ammonia surrogate allows the formation of the corresponding *N*-aryl imine adduct, which was then hydrolyzed to furnish the desired aniline **4t** with an overall acceptable yield of 45% (entry 20).

To broaden the substrate scope, we further investigated this MCR with respect to hydrazones 1. Gratifyingly, as depicted in Table 2 (entries 21–28), the entire coupling proceeded cleanly and selectively in good to excellent yields. Functionalized *N*-tosylhydrazones containing electrondonating or -withdrawing groups proved to be suitable substrates in this MCR. It is noteworthy that the chloro substituent on the phenyl ring of hydrazone 1c and 1d was tolerated, enabling further metal-catalyzed functionalization processes (entries 22–23).

The effectiveness of our protocol was also demonstrated with N-tosylhydrazones $\mathbf{1g}$, \mathbf{h} containing a secondary carbon atom α to the hydrazone function to provide tetrasubstituted olefins $\mathbf{4z}$ and $\mathbf{4aa}$ having a cycloalkylidene unit in good yields (entries 26 and 27). Finally, cyclic N-tosylhydrazone $\mathbf{1i}$, which is derived from cyclooctanone, also undergoes smooth MCR to afford $\mathbf{4ab}$ in 68% yield (entry 28).

As sulfonyl hydrazones could be simply prepared by mixing relative sulfonyl hydrazides and the carbonyl compounds, we investigated whether the reaction could be carried out in a one-pot fashion directly from carbonyl compounds 7, avoiding the isolation of the tosylhydrazone

intermediates 1. After some experimentation, we were pleased to successfully achieve a one-pot MCR protocol. In this process, carbonyl compounds **7a,b** and sulfonyl hydrazide were stirred in PhF for 2 h at 90 °C, and then all the reagents were added to the reaction mixture. Following this procedure, yields similar to those from hydrazones 1 were obtained (Scheme 2).

Scheme 2. One-Pot MCR from Carbonyl Compounds 7a,b

In conclusion, we have succeeded in developing a novel three-component reaction of N-tosylhydrazones, polyhaloarenes, and amines, allowing efficient sequential C=C bond formation and intermolecular C—N cross-coupling. Our optimized conditions allowed us to prepare 1,1diarylethylene derivatives 4 of biological interest that accommodated a wider combination of nitrogen substituents at the aromatic ring. Good to excellent yields were obtained using a wide range of amine partners, including primary and secondary anilines as well as aliphatic amines. The process is very general with regard to other coupling partners, the tosylhydrazone and the dihalogenated aromatic system. Moreover, it is not necessary to isolate the tosylhydrazone partner, and the reaction can be conducted in a one-pot manner directly from the carbonyl compound. All of these features make this simple and general MCR of broad utility for the synthesis and development of new medicinal agents. Further studies on the scope and synthetic applications are in progress.

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Supporting Information Available. Experimental procedures and spectroscopic data of new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

Org. Lett., Vol. 15, No. 1, 2013

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